

Chemistry Learning Journey

Beyond AGS

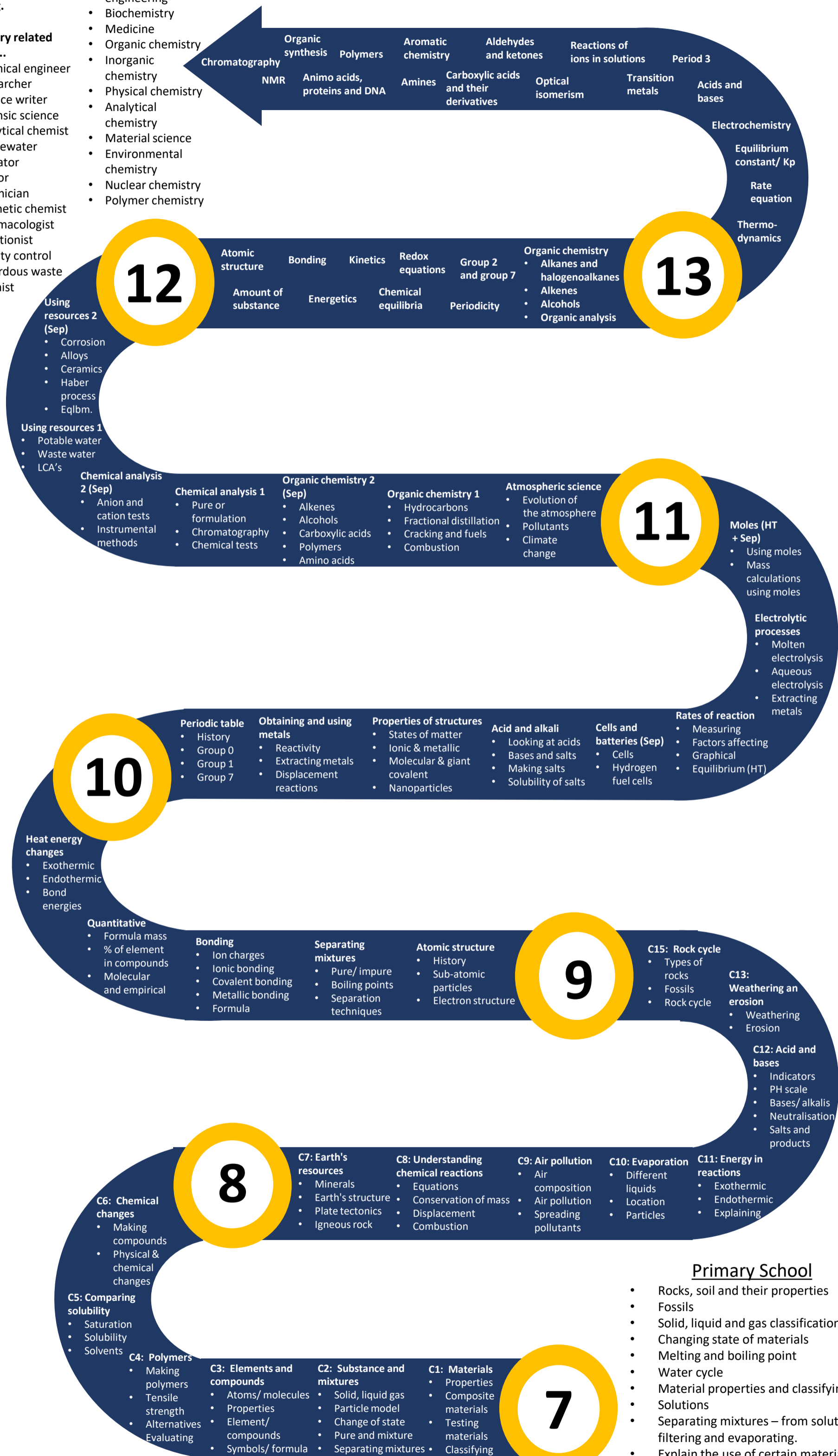
Or just continue to develop an interest in chemistry and develop life-long learning.

Chemistry related careers...

- Chemical engineer
- Researcher
- Science writer
- Forensic science
- Analytical chemist
- Wastewater operator
- Doctor
- Technician
- Synthetic chemist
- Pharmacologist
- Nutritionist
- Quality control
- Hazardous waste chemist

Continuing to study a chemistry related course...

- Chemical engineering
- Biochemistry
- Medicine
- Organic chemistry
- Inorganic chemistry
- Physical chemistry
- Analytical chemistry
- Material science
- Environmental chemistry
- Nuclear chemistry
- Polymer chemistry



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Atomic structure
Amount of substance

Bonding
Energetics

Kinetics
Chemical equilibria

Redox equations
Periodicity

Group 2 and group 7

Organic chemistry
• Alkanes and halogenoalkanes
• Alkenes
• Alcohols
• Organic analysis

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Electrochemistry
Equilibrium constant/ K_p
Rate equation
Thermodynamics

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Chemical analysis 2 (Sep)
• Anion and cation tests
• Instrumental methods

Chemical analysis 1
• Pure or formulation
• Chromatography
• Chemical tests

Organic chemistry 2 (Sep)
• Alkenes
• Alcohols
• Carboxylic acids
• Polymers
• Amino acids

Organic chemistry 1
• Hydrocarbons
• Fractional distillation
• Cracking and fuels
• Combustion

Atmospheric science
• Evolution of the atmosphere
• Pollutants
• Climate change

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Moles (HT + Sep)
• Using moles
• Mass calculations using moles

Electrolytic processes
• Molten electrolysis
• Aqueous electrolysis
• Extracting metals

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Periodic table
• History
• Group 0
• Group 1
• Group 7

Obtaining and using metals
• Reactivity
• Extracting metals
• Displacement reactions

Properties of structures
• States of matter
• Ionic & metallic
• Molecular & giant covalent
• Nanoparticles

Acid and alkali
• Looking at acids
• Bases and salts
• Making salts
• Solubility of salts

Cells and batteries (Sep)
• Cells
• Hydrogen fuel cells

Rates of reaction
• Measuring
• Factors affecting
• Graphical
• Equilibrium (HT)

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Heat energy changes
• Exothermic
• Endothermic
• Bond energies

Quantitative
• Formula mass
• % of element in compounds
• Molecular and empirical

Bonding
• Ion charges
• Ionic bonding
• Covalent bonding
• Metallic bonding
• Formula

Separating mixtures
• Pure/ impure
• Boiling points
• Separation techniques

Atomic structure
• History
• Sub-atomic particles
• Electron structure

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C15: Rock cycle
• Types of rocks
• Fossils
• Rock cycle

C13: Weathering and erosion
• Weathering
• Erosion

C12: Acid and bases
• Indicators
• PH scale
• Bases/ alkalis
• Neutralisation
• Salts and products

C6: Chemical changes
• Making compounds
• Physical & chemical changes

C5: Comparing solubility
• Saturation
• Solubility
• Solvents

C4: Polymers
• Making polymers
• Tensile strength
• Alternatives
• Evaluating

C3: Elements and compounds
• Atoms/ molecules
• Properties
• Element/ compounds
• Symbols/ formula

C2: Substance and mixtures
• Solid, liquid gas
• Particle model
• Change of state
• Pure and mixture
• Separating mixtures

C1: Materials
• Properties
• Composite materials
• Testing materials
• Classifying

C7: Earth's resources
• Minerals
• Earth's structure
• Plate tectonics
• Igneous rock

C8: Understanding chemical reactions
• Equations
• Conservation of mass
• Displacement
• Combustion

C9: Air pollution
• Air composition
• Air pollution
• Spreading pollutants

C10: Evaporation
• Different liquids
• Location
• Particles

C11: Energy in reactions
• Exothermic
• Endothermic
• Explaining

Primary School

- Rocks, soil and their properties
- Fossils
- Solid, liquid and gas classification
- Changing state of materials
- Melting and boiling point
- Water cycle
- Material properties and classifying
- Solutions
- Separating mixtures – from solution, filtering and evaporating.
- Explain the use of certain materials
- Reversible changes: dissolving, mixing and changes of state.
- Changes that make new materials; including burning and acid with bicarbonate of soda.